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Stereoselective generation of $cis-2$ -lithio-3-CF₃-oxirane via CF_3 -substituted β -oxido carbenoids. Highly stereoselective synthesis of CF_3 -substituted tri- and tetrasubstituted oxiranes and tetrasubstituted alkenes

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Abstract—Treatment of 2-substituted 3,3-dichloro-1,1,1-trifluoropropan-2-ols with organolithium reagents R^2 Li in THF at -98° C stereoselectively produces 2,3-disubstituted 2-lithio-3-trifluoromethyloxiranes with Li and CF₃ cis. The reagents react with electrophiles El-X or organoboranes R^3BR_2 to give CF₃-containing tri- and tetrasubstituted oxiranes or tetrasubstituted alkenes, respectively, with high diastereoselectivities.

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1. Introduction

Oxiranyllithiums have been emerging as highly versatile reagents for the stereocontrolled synthesis of functionalized oxiranes as well as alkenes.^{[1](#page-11-0)} Strategies for generation of oxiranyllithiums recorded so far are divided into two categories: one involves deprotonation of an oxirane with a base, and the other is exchange of heteroatoms such as sulfur and tin with lithium (Scheme 1). Both methods are synthetically valuable and widely utilized, while stereodefined oxiranes, whose preparations often require robust efforts, are essential as precursors for stereoselective generation of oxiranyllithiums.

On the other hand, intramolecular cyclization of β -oxido lithium carbenoids can, in principle, be the third strategy that is quite unique since acyclic precursors are employed in place of cyclic ones $(Scheme 1)$. β -Oxido carbenoids generated from dihalohydrins $(R^3=$ halogen) with a base, however, reportedly undergo 1,2-rearrangement of a substituent attached at the carbinol carbon, giving rise to carbonyl compounds.[2](#page-11-0) For instance, treatment of benzaldehyde with dichloromethyllithium in THF at -95° C followed by the addition of BuLi at -95° C generated phenyl-substituted β -oxide carbenoid in which, upon warming to 0° C, a phenyl group migrated in 1,2-fashion

Scheme 1. Generation of oxiranyllithiums.

without cyclization to produce the corresponding lithium enolate whose hydrolysis gave a homologated aldehyde in a good yield (Scheme 2).^{2e,h} Thus, intramolecular cyclization of dihalohydrins leading to oxiranyllithiums is still to be explored.^{[3](#page-11-0)}

Scheme 2. Rearrangement of B-oxido carbenoid.

Keywords: carbenoids; fluorine; lithium; oxirane; trifluoromethyl.

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In the course of our study on the novel synthesis of organofluorine compounds utilizing fluorine-containing carbenoid reagents, 4 we focused our attention on novel CF_3 -substituted β -oxido carbenoids and have found that the carbenoids are capable of the realization of the third strategy ([Scheme 1\)](#page-0-0).^{[5](#page-11-0)} We describe herein novel generation of oxiranyllithiums 3 from acyclic dichlorohydrins 1 and their reactions with an electrophile, providing convenient and versatile synthetic methods of preparing CF₃-containing tri- and tetrasubstituted oxiranes 4 and tetrasubstituted alkenes 5 and 6 (Scheme 3).[6](#page-11-0)

Scheme 3. Novel and stereoselective generation and reactions of CF_3 -substituted oxiranyllithiums 3. substituted oxiranyllithiums 3. Table 2. Reaction of 1a with BuLi $(R^2 = Bu)^a$

2. Results and discussion

2.1. Preparation of CF_3 -containing dichlorohydrins 1

Dichlorohydrins 1 were readily prepared by treatment of freshly distilled 3,3-dichloro-1,1,1-trifluoropropan-2-one with Grignard reagents R^1MgBr in THF at $\widehat{0}^{\circ}C$ (Eq. (1)). The results are summarized in Table 1. Various kinds of substituent $R¹$ were easily introduced in moderate to good yields. In contrast, use of the corresponding organolithium reagents $(R¹Li)$ resulted in the low yields of 1 probably because deprotonation of the ketone by R¹Li competed with the carbonyl addition.

ð1Þ

Table 1. Preparation of dichlorohydrins 1^a

R ¹		Yield $(\%)^b$
$Ph(CH_2)$	1a	59
(E) -PhCH=CH	1b	88
$PhC = C$	1c	86
Ph	1d	80

^a To a 1.0 M THF solution of $R¹MgBr$ (30 mmol) was added freshly distilled 3,3-dichloro-1,1,1-trifluoropropan-2-one (3.2 mL, 28 mmol) at 0 $^{\circ}$ C. The mixture was stirred for 10–30 min at 0 $^{\circ}$ C and stirred at room temperature overnight.
^b Isolated yields based on the ketone.

2.2. Synthesis of tri- and tetrasubstituted oxiranes 4

Whereas the rearrangement of β -oxido carbenoids generated from dihalohydrins is applicable to homologation of not only aldehydes and ketones but esters (via β -oxido alkylidene-type carbenoids), no precedent is available for CF_3 -substituted ketones. Hence, we became interested in a reaction of β -oxido carbenoids 2 to develop a novel method for synthesis of CF_3 -containing target compounds. Thus, we first treated 1a with 3 molar amounts of BuLi in THF at -78° C and then warmed the reaction mixture to room temperature before quenching with MeOH in a manner similar to the reported procedure^{[2e,h](#page-11-0)} (Eq. (2), $R^2 = Bu$). Unexpectedly, (E) -allylic alcohol 7 was obtained as the sole product in 58% yield (Table 2, entry 1). No products derived from the expected CF_3 or $Ph(CH_2)_2$ rearrangement were detected. When the same reaction was effected and quenched at -78° C, oxirane 4a was isolated in 33% yield along with 7 (58% yield, entry 2). Moreover, lowering the reaction temperature to -98° C gave 4a solely in 85% yield as a single diastereomer (entry 3).

^a To a solution of 1 (1.0 mmol) in THF (5 mL) was added R^2Li (3.0 mmol) at the temperature indicated. The solution was allowed to warm to room temperature (entry 1) or stirred at the temperature for 15 min before quenching with methanol (entries 2 and 3).

Under the same conditions, formation of oxirane of type 4 was general irrespective of substituents $R¹$ and $R²$ (Eq. (2) and Table 3). Thus, the reaction of dichlorohydrins 1b–d afforded the corresponding oxiranes 4b–d in good yields with good to high diastereoselectivity. In addition to BuLi,

Table 3. Synthesis of trisubstituted oxiranes 4^a

Run		R ¹	R^2		Yield $(\%)^b$	Isomer ratio ^c
$\overline{1}$	1b	(E) -PhCH=CH	Bu	4b	74	83:17
2	1c	$PhC = C$	Bu	4c	84	>95:55
3	1d	Ph	Bu	4d	83	89:11
$\overline{4}$	1a	$Ph(CH_2)$	Ph	4e	93	>95:55
5	1b	(E) -PhCH=CH	Ph	4f	87	>95: < 5
6	1c	$PhC = C$	Ph	4g	95	>95: < 5
7	1c	$PhC = C$	Me	4h	85	>95: < 5
8	1c	$PhC \equiv C$	C_2H_3	4i	67	>95 : < 5

^a To a solution of 1 (1.0 mmol) in THF (5 mL) was added R^2Li (3.0 mmol) at -98° C. The solution was stirred at -98° C for 15 min before quenching with methanol at -98° C. For entries 4–6, the reaction mixture was allowed to warm to -78° C before quenching with methanol at -78° C. Isolated yields.

 $\rm ^c$ Determined by $\rm ^1H$ and $\rm ^{19}F$ NMR analysis.

phenyllithium, methyllithium, and vinyllithium also reacted with $1a-c$ to give $4e-i$ stereoselectively. No reaction took place with phenylethynyllithium, while complex mixture resulted when s-BuLi or t-BuLi was employed with 1. In all cases, no product derived from the rearrangement was observed. A solvent mixture of THF/Et₂O/pentane $(4:1:1)$ was also effective for the production of 4, whereas use of such solvent(s) as Et_2O , $Et_2O/TMEDA$, hexane, and hexane/TMEDA resulted in the low conversion of 1 and/ or the low yield of 4. In addition, when 2 molar amounts of R2 Li were employed, 4 was produced in moderate yields with recovery of 1.

Since product distribution of 4a and 7 was dependent on the reaction temperature as shown in [Table 2](#page-1-0), we considered that oxiranyllithiums 3 should be produced as an intermediate at -98° C (for discussion of the mechanism, see Section 2.3). Accordingly, we treated 1 with BuLi or PhLi in THF at -98° C and then with an electrophile at -98° C. The resulting mixture was allowed to warm to -78° C and quenched with MeOH at -78° C. We found that silylation with chlorotrimethylsilane, borylation with (isopropoxy)- (pinacolato)borane, aldehyde addition, ketone addition, and allylation with allyl bromide proceeded smoothly to give the corresponding tetrasubstituted oxiranes 4j–r in good yields with excellent diastereoselectivity (Table 4). These results clearly demonstrate the generation of 3 from acyclic 1 and the realization of the third strategy in [Scheme 1.](#page-0-0)

In the case of alkylation, both diastereomers of tetrasubstituted oxiranes 4 can be prepared by proper choice of \mathbb{R}^2 Li and El-X. For example, treatment of 1c with BuLi and then iodomethane in HMPA afforded 4s, while another diastereomer $4s'$ was obtained by methyllithium-induced generation of 3 followed by trapping with iodobutane in HMPA as illustrated in Scheme 4.[7](#page-11-0)

Stereochemistry of 4 was determined by X-ray crystallographic analysis of benzophenone adducts 4o and 4q in which a CF_3 group and $Ph_2C(OH)$ were *cis* (Fig. 1).^{[8](#page-11-0)} These results clearly indicate that oxiranyllithiums 3 were

Scheme 4. Stereodivergent synthesis of tetrasubstituted oxiranes 4.

Table 4. Synthesis of tetrasubstituted oxiranes 4^a

Run	R^1	R^2	El ^b	4		Yield $(\%)^c$ Isomer ratio ^d
	$PhC = C$		Bu $Me3Si$	4j	79	>95: < 5
\overline{c}	$PhC \equiv C$	Bu	B(OCMe ₂) ₂	4k	62	>95: < 5
3	$PhC \equiv C$	Ph	B(OCMe ₂) ₂	-41	76	>95:55
4	$PhC \equiv C$	Bu	PhCH(OH)	4m	82	$^{\circ}$
5	$PhC \equiv C$	Bu	$Et_2C(OH)$	4n	71	>95:5
6	$PhC = C$	Bu	$Ph_2C(OH)$	40	68	>95: < 5
7	$Ph(CH_2)$	Bu	$Ph_2C(OH)$	4p	57	>95: < 5
8	(E) -PhCH=CH	Bu	$Ph_2C(OH)$	4q	55	89:11
9	$PhC = C$	Bu	C_3H_5	4r	70	>95: < 5

^a To a solution of 1 (1.0 mmol) in THF (5 mL) was added R^2Li (3 mmol) at -98° C and the mixture was stirred at -98° C for 15 min before adding El-X (1.5 mmol). The resulting solution was gradually warmed to

 -78° C and quenched with MeOH at -78° C.
^b El-X employed were as follows: entry 1, Me₃SiCl; entries 2 and 3, $(i-Prob)B(OCMe₂)$; entry 4, PhCHO; entry 5, Et₂CO; entries 6–8, Ph₂CO; entry 9, allyl bromide.

 σ ^c Isolated yield.
^d Determined by ¹H and ¹⁹F NMR analysis.

 $^{\circ}$ Two of four possible diastereomers were obtained in a ratio of 60:40.

stereoselectively generated as intermediates with $CF₃$ and Li being cis in all cases.^{[9](#page-11-0)}

 $4₀$

Figure 1. ORTEP drawings of 4o and 4q.

2.3. Mechanism for the generation of 3

The mechanism for the stereoselective generation of 3 from 1 is proposed in Scheme 5. At first, 1 would react with 2 molar amounts of R^2Li to produce β -oxido carbenoid 2. We assume that 2 favors a conformation in which a CF_3 group and a lithium atom connecting a carbenoid carbon are positioned synclinal due to lithium– fluorine chelation.^{[10](#page-11-0)} Starting from the Li–F chelated conformer 2, there are two plausible pathways. Substitution of chlorine in 2 with the third R^2Li from the OLi side would produce $2'$ which would undergo intramolecular cyclization to generate 3 that was stable at -98° C and reacted with an electrophile producing 4. Alternatively, cyclization of 2 would take place at first to generate chlorolithiooxirane $2ⁿ$ which might form oxonium ion $2^{\prime\prime\prime}$ by participation of a lone pair of electrons in oxygen. Subsequently, the third R^2Li would add to the carbon substituted by a lithium atom to afford 3 with retention of configuration of 2^{n} .^{[11](#page-11-0)} The fact that 3 was generated at some extent with remaining 1 by use of 2 molar amounts of R²Li indicated that deprotonation by

Scheme 5. Proposed mechanism for the stereoselective generation and reactions of oxiranyllithiums 3.

the third R²Li was a rate-determining step. At temperatures above -78° C, 3 would cause ring-opening via α elimination of lithium and oxygen to generate a carbene which would undergo adjacent C–H bond insertion with R^2 =Bu, giving rise to 7.

2.4. Reaction of β -MgO carbenoids leading to CF₃-containing tetrasubstituted alkenes 8

So far, we have demonstrated that β -LiO carbenoids react with an organolithium reagent to generate oxiranyllithiums 3 via intramolecular cyclization. We expected that changing lithium to magnesium might suppress the intramolecular cyclization and perform rearrangement reaction of β -oxido carbenoids. Thus, we treated 1a with MeMgBr in THF at -78° C to prepare the corresponding magnesium alkoxide. To the solution, we added 3 molar amount of MeLi at -78° C and warmed the mixture to room temperature (Scheme 6). No rearranged product, however, formed but tetrasubstituted alkene 8a was obtained in 60% yield (Table 5, entry 1). The alkene formation was applicable to 1b–d and BuLi as R^2 Li as shown in Table 5 although the yields of 8b and 8c were low.

2.5. Stereoselective synthesis of CF_3 -containing tetrasubstituted alkenes 5 and 6

Regio- and stereoselective synthesis of tetrasubstituted alkenes is a challenging issue in organic synthesis.^{[12](#page-11-0)} Oxiranyllithiums are reported to react with such organometallic reagents as RLi, R_3 Al, R_2Z n, and R_4 CeLi to give alkylated alkenes.^{[13](#page-11-0)} We envisioned that CF_3 -containing tetrasubstituted alkenes should be synthesized stereoselectively if such transformation would be applicable to 3. Then, we examined a reaction of 3 with various kinds of organometallic reagents. Treatment of 1c with more than 3 molar amounts of BuLi followed by warming to room

Scheme 6. Reaction of CF_3 -substituted β -MgO lithium carbenoids.

Table 5. Synthesis of tetrasubstituted alkenes 8^a

Entry		R'	R^2	8	Yield $(\%)^b$
1	1a	$Ph(CH_2)$	Me	8a	60
2	1b	(E) -PhCH=CH	Me	8b	15
3	1b	(E) -PhCH=CH	Bu	8с	26
4	1c	$PhC = C$	Me	8d	55
5	1c	$PhC = C$	Bu ^c	8e	71
6	1d	Ph	Me	8f	69
7	1d	Ph	Bu	8g	60

^a To a THF (3 mL) solution of 1 (1.0 mmol) was added R^2MgBr (1.0 mmol) at -78° C. After stirred at -78° C for 1 h, the solution was added R^2Li (3.0 mmol) at $-78^{\circ}C$ and was allowed to warm to room temperature before quenching with saturated aqueous NH₄Cl solution.
^b Isolated yields. c BuMgCl was used instead of BuMgBr.

Table 6. Stereoselective synthesis of tetrasubstituted alkenes 5 (R^1 = $PhC=Cl^a$

Entry	R^2	R^3		Yield $(\%)^b$	Isomer ratio
\overline{c}	Bu	$PhC \equiv C$	5c	74	>95 : <5
	Bu	(E) -BuCH=CH	5d	56	>95 : <5
3	Bu	Ph	5е	66	97:3
$\overline{4}$	Ph	Bu	5e'	84	96:4

 a To a solution of 1c (1.0 mmol) in THF (5 mL) was added BuLi (1.50 M, 2.0 mL, 3.0 mmol) at -98° C. The solution was stirred at -98° C for 15 min and then was added $R^3B(OCMe_2)_2$ (2.0 mmol) at $-98^{\circ}C$. The resulting solution was gradually warmed to room temperature and then refluxed for 4h before quenching with aqueous saturated NH₄Cl solution. **b** Isolated yield.

temperature and quenching with MeOH produced only 7, indicating that no reaction of $3(R^2=Bu)$ occurred.^{[14](#page-11-0)} When $3(R^2=Bu)$ was treated with Me₃Al or Et₂Zn, methylated or ethylated alkene 5a or 5b was obtained in low yields, respectively. On the other hand, triethylborane reacted with $3 \text{ (R}^2 = \text{Bu})$ to afford $5b$ in 45% yield with excellent stereoselectivity $(>95%)$.^{[15](#page-11-0)} Furthermore, (pinacolato)borane exhibited better reactivity (Eq. (3)). Thus, alkynyl, alkenyl, aryl, and alkyl groups were applicable to this stereospecific formation of 5 as a \mathbb{R}^3 substituent (Table 6). Noteworthy is that appropriate combination of \mathbb{R}^2 and \mathbb{R}^3 allows us to prepare either stereoisomer of 5 at will as demonstrated in entries 3 and 4.

Since a CF_3 and a phenyl groups in 5e were shown to be *cis* by X-ray analysis $(Fig. 2)$,^{[8](#page-11-0)} stereochemistry of 5 was assumed to be similar: CF_3 group and R^3 substituent are always cis.

Finally, reaction of 3 with bis(pinacolato)diboron and (dimethylphenylsilyl)(pinacolato)borane were studied to prepare β -CF₃-substituted alkenylboronates (Scheme 7).^{[16](#page-11-0)} Thus, treatment of a THF solution of $3(R^2=Bu)$ or Ph) with

Figure 2. ORTEP drawing of 5e.

R^2	Li_1 , Q_1 , CF_3 R^1 3		$B - B$ -98 °C to r.t.	B R^2 6E	CF ₃ $\ddot{}$ R'	R^2 CF ₃ R^1 В 6Z	
		R^2	Yield (%)	6Е	\vdots	6Z	
$B-Si$	-98 °C to r.t.	Bu Ph	76 86	98 >95		\overline{c} $<$ 5	
		Si R^2	CF_3 'n,	В R^2	CF ₃ $\ddot{}$ R ¹	R^2 CF ₃ 'n, в	
		9		6E		6Z	
R^2	Yield (%)	9		6E	:	6Z	
Bu Ph	54 45	30		7 7		93 63	

Scheme 7. Stereodivergent synthesis of β -CF₃-substituted alkenylboronates 6 (R^1 =PhC≡C, $B=$ B(OCMe₂)₂, Si=SiMe₂Ph).

the diboron at -98° C followed by warming the solution to room temperature gave stereoselectively $6E$ in which CF_3 and boryl group were arranged cis, respectively. On the other hand, alkenylboronates 6Z was stereoselectively produced in favor of $6E$ or β -CF₃-substituted alkenylsilane 9 when the silylborane was employed.

Stereoselective formation of $6E$ or $6Z$ can be explained by assuming a syn-elimination of LiOB or LiOSi at the last step as illustrated in [Scheme 8](#page-5-0). First, 3 should react with borane reagent R^3 -B to give borate 10 in which 1,2-migration of substituent $R³$ would take place and cause the ring-opening to generate lithium alkoxide 11. In cases of alkyl, alkenyl, alkynyl, aryl, and boryl groups for \mathbb{R}^3 , minimum rotation for eclipsing carbon–boron and carbon–oxygen bonds would lead to 12 from which 6E was produced by syn-elimination of LiOB.^{[17](#page-11-0)} When the silylborane was employed (i.e. R^3 in 11 was the silyl group), the carbon–boron bond should turn by 180° for LiOSi to eliminate in *cis* fashion, giving rise to $6Z^{18}$ $6Z^{18}$ $6Z^{18}$

3. Conclusion

We have demonstrated that CF_3 -substituted oxiranyllithiums can be generated from readily available acyclic CF3-containing dichlorohydrins and organolithium reagents via novel reaction of β -oxido carbenoids. Noteworthy is that the generation proceeds with high stereoselectivity that involves CF_3 and Li being *cis*. The oxiranyllithiums react stereospecifically not only with various kinds of electrophiles to give tri- and tetrasubstituted oxiranes but also with organoborane reagents to afford tetrasubstituted alkenes, which serve as highly versatile intermediates for stereocontrolled synthesis of CF_3 -containing complex organic molecules which are receiving much attention in pharma-ceutical and material sciences.^{[19](#page-11-0)}

Scheme 8. Proposed mechanism for stereodivergent formation of tetrasubstituted alkenes 6 ($B=B(OCMe₂)₂$, $Si=B_2Ph$).

4. Experimental

4.1. General remarks

Melting points were determined using a Yanagimoto Micro Melting Point Apparatus. ¹H NMR spectra were measured on a Varian Mercury 200 (200 MHz) spectrometer. The chemical shifts of ¹H NMR are expressed in parts per million downfield relative to the internal tetramethylsilane (δ =0 ppm) or chloroform (δ =7.26 ppm). Splitting patterns are indicated as s, singlet; d, doublet; t, triplet; q, quartet; brs, broad singlet. 13C NMR spectra were measured on a Varian Mercury 200 (50 MHz) spectrometer with tetramethylsilane as an internal standard (δ =0 ppm) or chloroform-d (δ =77.0 ppm). ¹⁹F NMR spectra were measured on a Varian Mercury 200 (188 MHz) spectrometer with CFCl3 as an internal standard (δ =0 ppm). Chemical shift values are given in parts per million downfield relative to the internal standard. Infrared spectra (IR) were recorded on a Shimadzu FTIR-8400 spectrometer. GC-MS analyses were performed with a JEOL JMS-700 spectrometer by electron ionization at 70 eV. Elemental analyses were carried out with a YANAKO MT2 CHN CORDER machine at Elemental Analysis Center of Kyoto University. TLC analyses were performed by means of Merck Kieselgel 60 F_{254} and column chromatography was carried out using Merck Kieselgel 60 (230–400 mesh). Preparative HPLC was carried out with a Japan Analytical Industry Co., Ltd, LC-908 chromatograph using a JAIGEL-1H and -2H GPC columns. Tetrahydrofuran was distilled from benzophenone and sodium before use under a nitrogen atmosphere. All reactions were carried out under an argon atmosphere.

4.2. Materials

Anhydrous tetrahydrofuran was purchased from Kanto Chemicals. Butyllithium in hexane, methyllithium in diethyl ether, phenyllithium in cyclohexane/diethyl ether were purchased from Aldrich or Kanto Chemicals, and titrated according to the literature method prior to use. 3,3- Dichloro-1,1,1-trifluoropropan-2-one was generously provided by Central Glass Co. Ltd.

4.3. General procedure for the preparation of dichlorohydrin 1

To a THF solution of R^1MgBr (30 mmol, 1.0 M) was added freshly distilled 3,3-dichloro-1,1,1-trifluoropropan-2-one $(3.2 \text{ mL}, 28 \text{ mmol})$ at 0°C . The mixture was stirred for $10-30$ min at 0° C and stirred at room temperature overnight. The reaction mixture was quenched with saturated aqueous $NH₄Cl$ solution and the aqueous layer was extracted with diethyl ether three times. The combined organic layer was dried over anhydrous $MgSO₄$ and concentrated in vacuo to give the crude compound, which was purified by column chromatography on silica gel (hexane/ethyl acetate 10:1) giving rise to 1.

4.3.1. 1,1-Dichloro-2-trifluoromethyl-4-phenylbutan-2 ol (1a). Yield: 59%. Colorless oil. R_f 0.22 (hexane/ethyl acetate 10:1). ¹H NMR δ 2.30-2.41 (m, 2H), 2.80 (t, J=8.6 Hz, 2H), 2.95 (brs, 1H), 5.99 (s, 1H), 7.21 - 7.37 (m, 5H). ¹³C NMR δ 28.6, 33.3, 72.9, 78.6 (q, J=26.8 Hz), 124.3 (q, $J=287.0$ Hz), 126.5, 128.3, 128.7, 140.2. ¹⁹F NMR (188 MHz, CDCl₃, δ -73.5. IR (neat): 3540, 3034, 1951, 1805, 1601, 1503, 1461, 1329, 1184, 1122, 911, 810, 782, 740, 671 cm⁻¹. MS m/z 290 (2, M⁺+4), 288 (9, M⁺+2), 287 (2, M⁺+1), 286 (14, M⁺), 91 (100). Anal. calcd for $C_{11}H_9C_7F_3O$: C, 46.02; H, 3.86. Found: C, 46.15; H, 3.80.

4.3.2. (E)-1,1-Dichloro-2-trifluoromethyl-4-phenyl-but-3-en-2-ol (1b). Addition reaction of styrylmagnesium bromide, prepared from a E/Z mixture of β -bromostyrene (Aldrich), to 3,3-dichloro-1,1,1-trifluoropropan-2-one gave a E/Z mixture of the corresponding adducts which was separated by silica gel column chromatography. Yield: 88%. Pale yellow oil. R_f 0.26 (hexane/ethyl acetate 10:1). ¹H NMR δ 3.07 (brs, 1H), 6.01 (s, 1H), 6.38 (d, J=15.9 Hz, 1H), 7.10 (d, J=15.9 Hz, 1H), 7.34–7.55 (m, 5H). ¹³C NMR δ 72.9, 79.0 (q, J=28.1 Hz), 119.0, 123.4 (q, J=286.1 Hz), 127.1, 128.8, 129.1, 134.9, 136.8. ¹⁹F NMR δ -75.3. IR (neat): 3540, 1646, 1489, 1450, 1182, 965, 744, 685 cm⁻¹. MS m/z 288 (0.6, M⁺+4), 286 (3, M⁺+2), 285 (0.7, $M^{+}+1$, 284 (5, M^{+}), 201 (100). Anal. calcd for $C_{11}H_9Cl_2F_3O$: C, 46.34; H, 3.18. Found: C, 46.51; H, 3.05.

4.3.3. 1,1-Dichloro-2-trifluoromethyl-4-phenylbut-3-yn-**2-ol (1c).** Yield: 86% . Pale yellow oil. R_f 0.13 (hexane/ethyl) acetate 10:1). ¹H NMR δ 3.53 (s, 1H), 5.96 (s, 1H), 7.37-7.43 (m, 3H), 7.52–7.57 (m, 2H). ¹³C NMR δ 71.8, 76.1 (q, $J=31.1$ Hz), 79.0, 89.9, 120.1, 122.2 (q, $J=285.7$ Hz), 128.5, 130.0, 132.3. ¹⁹F NMR δ -75.6. IR (neat): 3554, 2241, 1601, 1487, 1436, 1355, 1201, 1105, 982, 803, 753, 710, 685 cm⁻¹. MS m/z 286 (0.4, M⁺+4), 284 (3, M⁺+2), 283 (0.5, M^+ +1), 282 (4, M^+), 199 (100). Anal. calcd for $C_{11}H_7Cl_2F_3O$: C, 46.67; H, 2.49. Found: C, 46.69; H, 2.69.

4.3.4. 1,1-Dichloro-3,3,3-trifluoro-2-phenylpropan-2-ol (1d). Yield: 80% . Colorless oil. R_f 0.25 (hexane/ethyl acetate 10:1). ¹H NMR δ 3.50 (brs, 1H), 6.35 (s, 1H), 7.43– 7.47 (m, 3H), 7.57 – 7.62 (m, 2H). ¹³C NMR δ 74.0, 80.3 (q, $J=285.0$ Hz), 123.7 (q, $J=125.7$ Hz), 125.7, 128.7, 129.6, 134.0. ¹⁹F NMR δ -73.1. IR (neat): 3550, 1456, 1310, 1163, 1075, 1004, 930, 789, 702, 645 cm⁻¹. MS m/z 262 $(0.1, M⁺+4), 260 (0.5, M⁺+2), 259 (0.1, M⁺+1), 258 (1,$ M^+), 175 (100). Anal. calcd for C₉H₇Cl₂F₃O: C, 41.73; H, 2.72. Found: C, 42.02; H, 2.90.

4.4. General procedure for the preparation of trisubstituted oxiranes 4

To a solution of 1 (1.0 mmol) in THF (5 mL) was added $R²Li$ (3.0 mmol) at $-98^{\circ}C$. The solution was stirred at -98° C for 15 min before quenching with methanol. To the reaction mixture was added saturated aqueous NH4Cl solution, and the aqueous layer was extracted with diethyl ether three times. Combined organic layer was dried over anhydrous $MgSO₄$ and concentrated in vacuo to give a crude product. Purification by column chromatography on silica gel (hexane/diethyl ether 10:1) gave 4.

4.4.1. $(3R^*, 4S^*)$ -3,4-Epoxy-1-phenyl-3-trifluoromethyloctane (4a). Yield: 85% . Colorless oil. R_f 0.54 (hexane/ ethyl acetate 10:1). ¹H NMR δ 0.94 (t, J=6.8 Hz, 3H), $1.25-1.72$ (m, 6H), 2.07 (t, $J=8.6$ Hz, 2H), 2.86 (t, $J=8.6$ Hz, 1H), 3.26 (t, $J=6.0$ Hz, 1H), 7.21–7.38 (m, 5H). 13C NMR ^d 13.8, 22.4, 27.3, 28.2, 28.3, 30.7, 59.6 (q, $J=34.0$ Hz), 59.7, 124.4 (q, $J=278.3$ Hz), 126.3, 128.2, 128.6, 140.9. ¹⁹F NMR δ -75.4. IR (neat): 2951, 1767, 1601, 1446, 1332, 1259, 1148, 1021, 943, 905, 804, 741, 691 cm⁻¹. MS m/z 274 (0.1, M⁺+2), 273 (1, M⁺+1), 272 $(8, M⁺), 104 (100)$. Anal. calcd for C₁₅H₁₉F₃O: C, 66.16; H, 7.03. Found: C, 66.29; H, 6.96.

4.4.2. 1-Phenyl-3-trifluoromethyloct-4-en-3-ol (7). Yield: 58%. Colorless oil. R_f 0.28 (hexane/ethyl acetate 10:1). ¹H NMR δ 0.94 (t, J=7.4 Hz, 3H), 1.26–1.60 (m, 2H), 1.92– 2.22 (m, 5H), 2.70 (t, $J=13.4$ Hz, 2H), 5.57 (d, $J=15.6$ Hz, 1H), 6.01 (dt, J=15.6, 6.6 Hz, 1H), 7.19–7.38 (m, 5H). ¹³C NMR δ 15.1, 22.1, 28.6, 34.2, 35.6, 75.8 (q, J=27.2 Hz), 125.7, 126.1, 128.3, 128.4, 128.5, 132.6 (q, $J=298.0$ Hz), 141.4. ¹⁹F NMR δ -82.1. IR (neat): 3470, 2950, 1600, 1500, 1450, 1260, 1170, 980, 750, 700 cm⁻¹. MS m/z 274 $(0.3, M⁺+2), 273 (3, M⁺+1), 272 (19, M⁺), 91 (100).$ Anal. calcd for $C_{15}H_{19}F_3O$: C, 66.16; H, 7.03. Found: C, 66.39; H, 6.97.

4.4.3. $(3R^*, 4S^*)$ -3,4-Epoxy-1-phenyl-3-trifluoromethyl- (E) -oct-1-ene (4b). Yield: 74% of a mixture of both

diastereomers (83:17). Colorless oil. R_f 0.61 (hexane/ethyl acetate 10:1). Major $(3R^*, 4S^*)$ -diastereomer: ¹H NMR δ 0.90 (t, $J=7.0$ Hz, 3H), $1.35-1.62$ (m, 6H), 3.46 (t, $J=5.6$ Hz, 1H), 6.32 (d, $J=16.0$ Hz, 1H), 6.81 (d, J=16.0 Hz, 1H), 7.32–7.54 (m, 5H). ¹³C NMR δ 13.8, 22.2, 26.1, 27.9, 61.1 (q, $J=35.5$ Hz), 62.0, 115.4, 123.5 (q, J=277.6 Hz), 126.7, 128.67, 128.71, 133.0, 136.8. ¹⁹F NMR δ -75.9. IR (neat): 2949, 2848, 1652, 1491, 1441, 1322, 1263, 1151, 972, 935, 741, 984 cm⁻¹. MS m/z 272 (0.6, $M^{+}+2$), 271 (8, $M^{+}+1$), 270 (41, M^{+}), 213 (100, M^+ – CF₃). Anal. calcd for C₁₅H₁₇F₃O: C, 66.72; H, 6.32. Found: C, 66.65; H, 6.34. Minor $(3R^*A R^*)$ -diastereomer (assignable peaks only are shown): ¹H NMR δ 2.96 (t, $J=5.4$ Hz, 1H), 6.38 (d, $J=16.0$ Hz, 1H), 6.76 (d, J=16.0 Hz, 1H). ¹⁹F NMR δ -68.9.

4.4.4. $(3R^*, 4S^*)$ -3,4-Epoxy-1-phenyl-3-trifluoromethyloct-1-yne (4c). Yield: 84% . Colorless oil. R_f 0.59 (hexane/ethyl acetate 10:1). ¹H NMR δ 0.95 (t, J=7.2 Hz, 3H), 1.37–1.63 (m, 4H), 1.79–1.90 (m, 2H), 3.44 (t, $J=6.0$ Hz, 1H), 7.34–7.42 (m, 3H), 7.49–7.54 (m, 2H). ¹³C NMR δ 13.9, 22.3, 27.7, 28.6, 61.5, 54.0 (q, J=40.4 Hz), 78.1, 88.1, 120.8, 122.0 (q, J=276.2 Hz), 128.4, 129.6, 132.2. ¹⁹F NMR δ -75.8. IR (neat): 2951, 2861, 2242, 1487, 1458, 1346, 1315, 1211, 1193, 1162, 1100, 1002, 924, 754, 721, 690 cm⁻¹. MS m/z 270 (0.3, M⁺+2), 269 (2, M^{+} +1), 268 (11, M⁺), 199 (4, M⁺ - CF₃), 182 (100). Anal. calcd for $C_{15}H_{15}F_3O$: C, 67.16; H, 5.64. Found: C, 66.98; H, 5.71.

4.4.5. $(1R^*$, $2S^*)$ -1, 2 -Epoxy-1-phenyl-1-trifluoromethylhexane (4d). Yield: 83% of a mixture of both diastereomers (89:11). Colorless oil. R_f 0.59 (hexane/ethyl acetate 10:1). Major $(1R^*, 2S^*)$ -diastereomer: ¹H NMR δ 0.82 (t, $J=7.0$ Hz, 3H), $1.02-1.63$ (m, 6H), 3.51 (t, $J=7.0$ Hz, 1H), 7.39–7.46 (m, 5H). 13C NMR ^d 13.7, 22.2, 27.7, 27.9, 60.7, 62.0 (q, $J=35.5$ Hz), 123.6 (q, $J=277.3$ Hz), 126.7, 128.4, 129.0, 129.7. ¹⁹F NMR δ -75.0. IR (neat): 2951, 2925, 2860, 1756, 1678, 1451, 1179, 1068, 903, 762, 721, 701 cm⁻¹. MS m/z 246 (1, M⁺+2), 245 (13, M⁺+1), 244 (100, M⁺), 187 (59, M⁺-Bu), 175 (55, M⁺-CF₃). Anal. calcd for $C_{13}H_{15}F_3O$: C, 63.93; H, 6.19. Found: C, 63.82; H, 6.09. Minor $(1R^*, 2R^*)$ -diastereomer (only assignable peaks are shown): ¹H NMR δ 2.94–3.01 (m, 1H). ¹⁹F NMR δ $-66.7.$

4.4.6. $(1R^*2S^*)$ -1,2-Epoxy-1,4-diphenyl-2-trifluoromethylbutane (4e). Yield: 93%. Colorless oil. R_f 0.58 (hexane/ethyl acetate 10:1). ¹H NMR δ 1.84 (t, J=8.6 Hz, 2H), 2.69 (m, 2H), 4.41 (s, 1H), 6.88 (m, 2H), 7.12–7.25 (m, 3H), 7.30–7.42 (m, 5H). 13C NMR ^d 27.4, 30.5, 59.7, 61.7 $(q, J=34.3 \text{ Hz})$, 123.0 $(q, J=279.2 \text{ Hz})$, 126.1, 126.4, 128.1, 128.4, 128.5, 128.7, 132.6, 140.7. ¹⁹F NMR δ -75.1. IR (neat): 3064, 3028, 2939, 2871, 1604, 1496, 1456, 1344, 1168, 1147, 1037, 941, 918, 871, 698 cm⁻¹. MS m/z 293 $(0.8, M⁺+1), 292 (5, M⁺), 201 (100).$ Anal. calcd for $C_{17}H_{15}F_3O$: C, 69.85; H, 5.17. Found: C, 69.85; H, 5.25.

4.4.7. (E) - $(3R^*$,4S $*)$ -3,4-Epoxy-1,4-phenyl-3-trifluoromethylbut-1-ene (4f). Yield: 87% as a colorless solid, mp 51–52°C. R_f 0.58 (hexane/ethyl acetate 10:1). ¹H NMR δ 4.57 (s, 1H), 5.99 (d, J=16.4 Hz, 1H), 6.87 (d, J=16.4 Hz, 1H), $7.25 - 7.32$ (m, 5H). ¹³C NMR δ 61.5, 62.7 (q,

J=35.3 Hz), 114.5, 122.0 (q, J=278.1 Hz), 126.8, 127.0, 128.2, 128.58, 128.62, 128.7, 135.2, 138.4. ¹⁹F NMR δ -75.3 . MS m/z 291 (6, M⁺+1), 290 (33, M⁺), 289 (3, M^+ – 1), 105 (100). Anal. calcd for C₁₇H₁₃F₃O: C, 70.34; H, 4.51. Found: C, 70.26; H, 4.46.

4.4.8. $(3R^*$, $4S^*$)-3, 4-Epoxy-1, 4-diphenyl-3-trifluoromethylbut-1-yne (4g). Yield: 95%. Colorless oil. R_f 0.50 (hexane/ethyl acetate 10:1). ¹H NMR δ 4.51 (s, 1H), 7.27-7.36 (m, 4H), 7.37 – 7.52 (m, 6H). ¹³C NMR (CD₃COCD₃) δ 57.4 (q, J=40.2 Hz), 62.8, 78.5, 89.9, 121.2, 123.0 (q, J=276.2 Hz), 128.5, 129.2, 129.7, 130.5, 131.0, 132.6, 132.9. ¹⁹F NMR δ -75.6. IR (neat): 3067, 3038, 2233, 1491, 1458, 1406, 1356, 1327, 1213, 1184, 1128, 1013, 916, 889, 756, 721, 696 cm⁻¹. MS m/z 288 (0.1, M⁺+2), 289 (3, M^+ +1), 288 (18, M⁺), 219 (11, M⁺ - CF₃), 182 (100). Anal. calcd for $C_{17}H_{11}F_3O$: C, 70.83; H, 3.85. Found: C, 71.12; H, 3.96.

4.4.9. $(3R^*, 4S^*)$ -3,4-Epoxy-1-phenyl-3-trifluoromethylpent-1-yne (4h). Yield: 85% . Colorless oil. R_f 0.49 (hexane/ethyl acetate 10:1). ¹H NMR δ 1.57 (d, J=5.2 Hz, 3H), 3.57 (q, J=5.2 Hz, 1H), 7.27-7.37 (m, 3H), 7.41-7.55 (m, 2H). ¹³C NMR δ 14.4, 54.3 (g, J=40.2 Hz), 57.6, 77.9, 88.3, 120.7, 121.9 (q, J=276.3 Hz), 128.4, 129.6, 132.2. ¹⁹F NMR δ -75.8. IR (neat): 3152, 3002, 2925, 2225, 1606, 1491, 1442, 1409, 1355, 1305, 1209, 1178, 1056, 911, 755, 695 cm⁻¹. MS m/z 227 (2, M⁺+1), 226 (14, M⁺), 182 (100). Anal. calcd for $C_{12}H_9F_3O$; C, 63.72; H, 4.01. Found: C, 64.00; H, 4.26.

4.4.10. $(3R^*, 4S^*)$ -3,4-Epoxy-1-phenyl-3-trifluoromethylhex-5-en-1-yne (4i). Yield: 67% . Colorless oil. R_f 0.50 (hexane/ethyl acetate 10:1). ¹H NMR δ 3.89 (d, $J=7.0$ Hz, 1H), $5.59-5.95$ (m, 3H), $7.29-7.46$ (m, 3H), 7.49–7.59 (m, 2H). ¹³C NMR δ 55.2 (q, J=40.6 Hz), 61.2, 78.7, 88.7, 120.6, 121.6 (q, J=276.8 Hz), 124.8, 128.4, 129.7, 130.1, 132.1. ¹⁹F NMR δ -75.9. IR (neat): 3061, 2928, 2241, 1491, 1445, 1356, 1298, 1221, 1182, 1153, 1130, 1009, 982, 957, 908, 756, 729, 689 cm⁻¹. MS m/z 240 $(0.3, M⁺+2), 239 (4, M⁺+1), 238 (31, M⁺), 169 (48,$ M^+ – CF₃), 141 (100). Anal. calcd for C₁₃H₉F₃O: C, 65.55; H, 3.81. Found: C, 65.50; H, 4.09.

4.4.11. $(3R^*, 4R^*)$ -3,4-Epoxy-1-phenyl-3-trifluoromethyl-4-(trimethylsilyl)oct-1-yne (4j). Purified by silica gel column chromatography (hexane/diethyl ether 10:1) followed by GPC. Yield: 79%. Colorless oil. R_f 0.63 (hexane/ethyl acetate 10:1). ¹H NMR δ 0.24 (s, 9H), 0.94 (t, $J=7.1$ Hz, 3H), $1.31-1.71$ (m, 5H), $2.05-2.17$ (m, 1H), 7.29–7.40 (m, 3H), 7.44–7.51 (m, 2H). ¹³C NMR δ 1.22 (q, $J=2.7$ Hz), 13.9, 23.0, 27.4, 36.2, 58.5 (q, $J=42.9$ Hz), 65.3, 81.0, 86.7, 120.2, 123.0 (q, J=276.6 Hz), 128.4, 129.3, 132.0. ¹⁹F NMR δ -70.7. IR (neat): 3049, 2951, 2860, 2224, 1598, 1488, 1335, 1250, 1181, 1069, 1020, 908, 842, 755, 686 cm⁻¹. MS m/z 342 (0.3, M⁺+2), 341 (1, M⁺+1), 340 (5.0, M⁺), 271 (100, M⁺-CF₃). Anal. calcd for $C_{18}H_{23}F_3OSi$: C, 63.50; H, 6.81. Found: C, 63.43; H, 6.76.

4.4.12. $(3R^*$, $4S^*$)-3, 4-Epoxy-1-phenyl-3-trifluoromethyl-4-(4,4,5,5-tetramethyl-1,3,2-dioxaborolan-2-yl) oct-1-yne (4k). Yield: 62% . Pale yellow oil. R_f 0.38 (hexane/ethyl acetate 10:1). ¹H NMR (500 MHz) δ 0.93

(t, J¼7.3 Hz, 3H), 1.30 (s, 12H), 1.17–1.60 (m, 4H), 1.81 (m, 1H), 1.98 (m, 1H), 7.31–7.37 (m, 3H), 7.46–7.48 (m, 2H). 13C NMR (125 MHz) ^d 13.8, 22.6, 24.5, 24.6, 27.8, 32.9, 57.2 (q, J=40.2 Hz), 60.6, 79.2, 84.9, 88.1, 120.9, 121.2 (q, J=276.2 Hz), 128.3, 129.4, 132.0. ¹⁹F NMR δ -73.2 . IR (neat): 3060, 2979, 2873, 2235, 1444, 1340, 1188, 1134, 1033, 850, 758 cm⁻¹. MS m/z 394 (4, M⁺), 393 (16, M⁺-1), 310 (100). Anal. calcd for C₂₁H₂₆BF₃O₃: C, 63.98; H, 6.65. Found: C, 64.07; H, 6.42.

4.4.13. $(3R^*$,4S $*)$ -3,4-Epoxy-1,4-diphenyl-3-trifluoromethyl-4-(4,4,5,5-tetramethyl-1,3,2-dioxaborolan-2 yl)but-1-yne (4l). Yield: 76% as an off-white solid, mp 82° C (dec). R_f 0.35 (hexane/ethyl acetate 10:1). ¹H NMR (500 MHz) δ 1.25 (s, 6H), 1.26 (s, 6H), 7.02–7.04 (m, 2H), 7.14–7.18 (m, 2H), 7.22–7.26 (m, 1H), 7.30–7.38 (m, 3H), 7.54–7.57 (m, 2H). ¹³C NMR (125 MHz) δ 24.3, 24.6, 59.1 $(q, J=40.2 \text{ Hz})$, 61.6, 78.8, 85.2, 88.3, 120.6, 121.0 $(q,$ J=276.2 Hz), 126.5, 127.9, 128.1 (2C), 129.3, 131.9, 134.8. ¹⁹F NMR δ -73.2. MS *mlz* 415 (4, M⁺+1), 414 (16, M⁺), 413 (16, M⁺-1), 331 (100). Anal. calcd for $C_{23}H_{22}BF_3O_3$: C, 66.69; H, 5.35. Found: C, 66.65; H, 5.54.

4.4.14. $(3R^*$, $4R^*$)-3,4-Epoxy-4-(hydroxyphenyl)methyl-1-phenyl-3-trifluoromethyloct-1-yne (4m). Purified by silica gel column chromatography (hexane/diethyl ether 10:1) followed by GPC. Yield: 82% of a 60:40 diastereomeric mixture. Pale yellow oil. R_f 0.35 (hexane/ethyl acetate 10:1). Major $(3R^*A R^*)$ -diastereomer: ¹H NMR δ 0.77 (t, $J=7.2$ Hz, 3H), 0.92–1.28 (m, 3H), 1.39–1.55 (m, 1H), 1.65–1.88 (m, 1H), 1.93–2.08 (m, 1H), 2.40 (brs, 1H), 5.14 (s, 1H), 7.30–7.52 (m, 10H). ¹³C NMR δ 13.7, 23.0, 27.4, 29.0, 58.8 (q, $J=40.2$ Hz), 72.2 (q, $J=4.2$ Hz), 73.2, 79.3, 89.0, 120.7, 122.6 (q, J=277.9 Hz), 125.6, 127.9, 128.3, 128.5, 129.6, 132.0, 138.3. ¹⁹F NMR δ -66.5. IR (neat): 3449, 3065, 2961, 2934, 2874, 2235, 1603, 1493, 1452, 1381, 1325, 1246, 1182, 1055, 937, 841, 756, 731, 689 cm⁻¹. MS m/z 375 (0.1, M⁺+1), 374 (0.4, M⁺), 176 (90), 183 (45), 133 (100). Anal. calcd for $C_{22}H_{21}F_3O_2$: C, 70.58; H, 5.65. Found: C, 70.46; H, 5.62. Minor $(3R^*, 4S^*)$ diastereomer (only assignable peaks are shown): ¹H NMR δ 0.72 (t, J=7.8 Hz, 3H), 2.41 (brs, 1H), 5.04 (s, 1H). ¹³C NMR δ 13.6, 22.8, 26.4, 29.0, 60.4 (q, J=40.0 Hz), 72.7 (q, $J=3.4$ Hz), 71.4, 79.7, 88.7, 120.9, 122.7 (q, $J=278.0$ Hz), 125.6, 128.0, 128.4, 128.5, 129.5, 132.0, 139.6. ¹⁹F NMR δ $-65.8.$

4.4.15. $(3R^*$, $4R^*$)-3, 4-Epoxy-4-(diethylhydroxy) methyl-1-phenyl-3-trifluoromethyloct-1-yne (4n). Purified by silica gel column chromatography (hexane/diethyl ether 10:1) followed by GPC. Yield: 71% . Yellow oil. R_f 0.38 (hexane/ethyl acetate 10:1). ¹H NMR δ 0.93 (t, J=7.0 Hz, 3H), 0.99 (t, J=7.6 Hz, 6H), 1.23-1.57 (m, 5H), 1.64-1.82 (m, 4H), 1.88 (brs, 1H), 2.05–2.28 (m, 1H), 7.24– 7.39 (m, 3H), 7.44–7.50 (m, 2H). ¹³C NMR δ 7.45, 7.78, 13.9, 23.1, 27.3, 29.5, 30.8, 31.6, 58.5 (q, $J=42.1$ Hz), 74.7, 76.8, 81.3, 89.1, 121.2, 123.1 (q, $J=277.4$ Hz), 128.4, 129.4, 131.9. ¹⁹F NMR δ -60.0. IR (neat): 3601, 2963, 2858, 1485, 1459, 1387, 1305, 1240, 1183, 1150, 1068, 1153, 1070, 1038, 911, 838, 746, 685 cm⁻¹. MS m/z 356 (4, M⁺+2), 355 (23, M⁺+1), 354 (100, M⁺), 183 (89). Anal. calcd for $C_{20}H_{25}F_3O$: C, 67.78; H, 7.11. Found: C, 68.06; H, 7.00.

4.4.16. $(3R^*$,4R $*)$ -3,4-Epoxy-4-(hydroxydiphenyl)methyl-1-phenyl-3-trifluoromethyloct-1-yne (4o). Purified by silica gel column chromatography (hexane/diethyl ether 10:1) followed by GPC. Yield: 68% as colorless crystals, mp 127°C. R_f 0.32 (hexane/ethyl acetate 10:1). ¹H NMR δ 0.29–0.47 (m, 1H), 0.52 (t, J=7.2 Hz, 3H), 0.79– 0.94 (m, 2H), $0.98-1.26$ (m, 1H), 1.83 (dt, $J=12.6$, 4.8 Hz, 1H), 2.26 (dt, $J=12.6$, 4.8 Hz, 1H), 3.23 (brs, 1H), 7.24– 7.52 (m, 15H). 13C NMR ^d 13.5, 22.3, 26.0, 33.1, 60.4 (q, J=42.2 Hz), 75.6, 80.1 (q, J=278.3 Hz), 80.7, 90.1, 120.9, 121.8 (q, J=278.3 Hz), 127.8, 127.9, 128.05, 128.09 (2C), 128.2, 128.5, 129.6, 132.0, 140.8, 141.7, ¹⁹F NMR δ –60.6. IR (nujol): 3563, 2920, 2850, 2230, 1493, 1310, 1238, 1193, 1180, 960, 760, 698, 660 cm⁻¹. MS m/z 452 (0.1, M⁺+2), 451 (3, M⁺+1), 450 (10, M⁺), 393 (21, M⁺-Bu), 183 (100), 105 (80). Anal. calcd for $C_{28}H_{25}F_3O$: C, 74.65; H, 5.59. Found: C, 74.85; H, 5.68.

4.4.17. $(3R^*$, $4R^*)$ -3, 4-Epoxy-4-(hydroxydiphenyl)methyl-1-phenyl-3-trifluoromethyloctane (4p). Purified by silica gel column chromatography (hexane/diethyl ether 10:1) followed by GPC. Yield: 57% as colorless crystals, mp 117°C. R_f 0.30 (hexane/ethyl acetate 10:1). ¹H NMR δ 0.11–0.22 (m, 1H), 0.52 (t, J=7.0 Hz, 3H), 0.66– 0.98 (m, 4H), 1.79–1.96 (m, 2H), 2.23–2.38 (m, 1H), 2.96 (t, J¼9.6 Hz, 2H), 3.40 (brs, 1H), 7.18–7.45 (m, 15H). 13C NMR δ 13.4, 22.2, 26.5, 30.4, 31.4, 33.3, 66.2 (q, $J=39.4$ Hz), 73.2, 80.1, 124.1 (q, $J=281.5$ Hz), 126.5, 127.6 (2C), 127.9, 128.0 (2C), 128.11, 128.13, 128.7, 140.7, 142.1. ¹⁹F NMR δ -59.7. IR (nujol): 3568, 2924, 2855, 1493, 1305, 1259, 1173, 1155, 1107, 1043, 883, 772, 706 cm⁻¹. MS m/z 455 (0.1, M⁺+1), 454 (0.2, M⁺), 183 (100), 105 (85). Anal. calcd for $C_{28}H_{29}F_3O$: C, 73.99; H, 6.43. Found: C, 74.09; H, 6.59.

4.4.18. $(3R^*A R^*)-3A-Bpoxy-4-(hydroxydipheny)$ methyl-1-phenyl-3-trifluoromethyloct-1-ene (4q). Purified by silica gel column chromatography (hexane/diethyl ether 10:1) followed by GPC. Yield: 55% of a diastereomeric mixture of 89: 11. Major $(3R^*, 4R^*)$ -isomer was separated from the mixture in 46% yield as colorless crystals, mp. 124°C. R_f 0.29 (hexane/ethyl acetate 10:1). ¹H NMR δ 0.09–0.26 (m, 1H), 0.38 (t, J=7.0 Hz, 3H), 0.48– 0.88 (m, 3H), 1.69 (dt, $J=10.8$, 5.4 Hz, 2H), 3.47 (brs, 1H), 6.44 (d, J=16.0 Hz, 1H), 6.84 (d, J=16.0 Hz, 1H), 7.27– 7.48 (m, 15H). ¹³C NMR δ 13.3, 22.2, 26.6, 31.2, 67.6 (q, $J=37.2$ Hz), 75.5, 79.9, 119.8, 123.3 (q, $J=280.0$ Hz), 126.8, 127.6, 127.7 (2C), 127.9, 128.1, 128.3, 128.6, 128.8, 134.1, 142.7, 145.5. ¹⁹F NMR δ -60.7. IR (nujol): 3574, 2924, 2852, 1493, 1448, 1254, 1190, 1146, 1115, 1086, 1026, 974, 914, 764, 752, 700, 665 cm⁻¹. MS m/z 454 (0.2, M^{+} +2) 453 (4, M^{+} +1), 452 (13, M^{+}), 252 (92), 183 (69), 105 (100). Anal. calcd for $C_{28}H_{27}F_3O$: C, 74.32; H, 6.01. Found: C, 74.08; H, 5.91.

4.4.19. $(3R^*A R^*)-3,4-Epoxy-4-(prop-1-en-3-yl)-1-ph$ enyl-3-trifluoromethyloct-1-yne (4r). To a solution of 1c (1.0 mmol) in THF (5 mL) was added BuLi (3.0 mmol) at -98° C. The solution was stirred at -98° C for 15 min. Allyl bromide $(130 \mu L, 1.5 \text{ mmol})$ in hexamethylphosphoric triamide (HMPA) (1.0 mL, 6 mmol) was added to the solution at -98° C and the resulting solution was allowed to warm gradually to -78° C over a period of 3 h before quenching with saturated aqueous $NH₄Cl$ solution. Usual workup and silica gel column chromatography (hexane/ diethyl ether 10: 1) followed by GPC gave 4r. Yield: 70%. Pale yellow oil. R_f 0.50 (hexane/ethyl acetate 10:1). ¹H NMR δ 0.93 (t, J=7.1 Hz, 3H), 1.33–1.60 (m, 4H), 1.73– 2.07 (m, 2H), $2.48 - 2.70$ (m, 2H), 5.20 (d, $J=11.2$ Hz, 1H), 5.22 (d, $J=15.8$ Hz, 1H), $5.73-5.94$ (m, 1H), $7.29-7.39$ (m, 3H), 7.42–7.49 (m, 2H). ¹³C NMR δ 13.9, 22.7, 26.7, 34.0 $(q, J=3.1 \text{ Hz})$, 58.6 $(q, J=34.8 \text{ Hz})$, 69.7, 80.2, 87.9, 119.0, 121.1, 122.6 (q, J=277.5 Hz), 128.4, 129.5, 131.9, 132.1. ¹⁹F NMR δ -67.2. IR (neat): 3084, 2961, 2874, 2235, 1643, 1491, 1445, 1406, 1333, 1250, 1184, 1155, 1049, 924, 756, 727, 689 cm⁻¹. MS m/z 309 (0.2, M⁺+1), 308 (1, M⁺), 182 (100). Anal. calcd for $C_{18}H_{19}F_3O$: C, 70.12; H, 6.21. Found: C, 70.39; H, 6.22.

4.4.20. $(3R^*A R^*)$ -4-(Dimethylphenyl)silyl-3,4-epoxy-1phenyl-3-trifluoromethyloct-1-yne (4s). To a solution of 1c (280 mg, 1.0 mmol) in THF (5 mL) was added BuLi $(1.50 \text{ M}, 2.0 \text{ mL}, 3.0 \text{ mmol})$ at -98°C . The solution was stirred at -98° C for 15 min. Iodomethane (93 μ L, 1.5 mmol) in HMPA (1.0 mL, 6.0 mmol) were added to the solution at -98° C and the resulting solution was gradually warmed to -78° C over a period of 3 h before quenching with aqueous saturated $NH₄Cl$ solution. Usual workup followed purification on silica gel column chromatography (hexane/diethyl ether 10:1) gave 4s (180 mg, 64% yield). Colorless oil. R_f 0.58 (hexane/ethyl acetate 10:1). ¹H NMR δ 0.46 (s, 3H), 0.52 (s, 3H), 0.71 (t, J=7.2 Hz, 3H), 0.89–1.62 (m, 5H), 1.90 (dt, $J=12.2$, 4.5 Hz, 1H), 7.29– 7.64 (m, 10H). ¹³C NMR δ -2.78, -3.29, 13.5, 22.7, 26.8, 36.3, 58.7 (q, J=41.0 Hz), 65.8, 80.9, 86.9, 121.2, 122.9 (q, J=277.0 Hz), 127.7, 128.4, 129.3, 129.5, 132.0, 134.2, 136.5. ¹⁹F NMR δ -70.4. IR (neat): 30.61, 2950, 2859, 2221, 1483, 1419, 1332, 1252, 1180, 1162, 1098, 1071, 1118, 903, 810, 753, 693 cm⁻¹. MS m/z 404 (3, M⁺+2), 403 (11, M^{+} +1), 402 (36, M^{+}), 135 (100). Anal. Calcd for $C_{23}H_{25}F_3OSi$: C, 68.63; H, 6.26. Found: C, 68.49; H, 6.25.

4.4.21. $(3R^*, 4R^*)$ -3,4-Epoxy-4-methyl-1-phenyl-3-tri**fluoromethyloct-1-yne (4s').** Yield: 77%. Colorless oil. R_f 0.58 (hexane/ethyl acetate 10:1). ¹H NMR δ 0.93 (t, $J=7.0$ Hz, 3H), $1.26-1.52$ (m, 4H), 1.62 (s, 3H), $1.75-$ 1.86 (m, 2H), 7.28–7.42 (m, 3H), 7.47–7.53 (m, 2H). 13C NMR δ 13.8, 20.9, 22.7, 32.3 (q, J=3.1 Hz), 58.6 (q, $J=39.7$ Hz), 68.3, 80.5, 87.8, 121.1, 122.7 (q, $J=277.4$ Hz), 128.4, 129.4, 132.1. ¹⁹F NMR δ -67.7. IR (neat): 3059, 2961, 2936, 2874, 2235, 1491, 1445, 1385, 1331, 1252, 1188, 1155, 1063, 1011, 922, 897, 756, 729, 691 cm⁻¹. MS m/z 284 (0.2, M⁺+2), 283 (1, M⁺+1), 282 (7, M⁺), 267 (1, M^+ – Me), 182 (100). Anal. calcd for C₁₆H₁₇F₃O: C, 68.07; H, 6.07. Found: C, 68.02; H, 6.04.

4.5. General procedure for the preparation of tetrasubstituted alkenes 8

To a solution of 1 (1.0 mmol) in THF (3 mL) was added Grignard reagent R^2MgBr (1.0 mmol) at $-78^{\circ}C$. After stirred at -78° C for 1 h, the solution was added R²Li (3.0 mmol) at -78° C and was allowed to warm to room temperature gradually before quenching with saturated aqueous NH4Cl solution. The aqueous layer was extracted with diethyl ether three times and the combined organic layer was dried over anhydrous $MgSO₄$ and concentrated in vacuo. The crude residue was purified by column chromatography on silica gel giving rise to 8.

4.5.1. 2-Methyl-5-phenyl-3-trifluoromethylpet-2-ene (8a). Yield: 60%. Colorless oil. R_f 0.63 (hexane/ethyl acetate 10:1). ¹H NMR δ 1.63 (q, J=2.2 Hz, 3H), 1.88 (q, $J=2.6$ Hz, 3H), 2.47 (t, $J=8.6$ Hz, 2H), 2.69 (m, 2H), 7.18– 7.36 (m, 5H). ¹³C NMR δ 21.5, 21.6, 30.6 (q, J=2.7 Hz), 35.3, 123.7 (q, J=27.0 Hz), 125.2 (q, J=274.3 Hz), 126.0, 128.4, 128.5, 141.3, 141.4, ¹⁹F NMR δ -57.8. IR (neat): 3030, 2930, 1660, 1600, 1600, 1500, 1450, 1320, 1220, 1180, 1110, 1040, 750, 700 cm⁻¹. MS m/z 230 (0.2, $M^{+}+2$), 229 (3, $M^{+}+1$), 228 (18, M^{+}), 91 (100). HRMS (EI) m/z M⁺ calcd for C₁₃H₁₅F₃, 228.1126; found, 228.1109. Anal. calcd for $C_{13}H_{15}F_3$: C, 68.41; H, 6.62. Found: C, 68.67; H, 6.90.

4.5.2. 1-Phenyl-4-methyl-3-trifluoromethylpenta-1,3 diene (8b). Yield: 15%. Colorless oil. R_f 0.50 (hexane/ ethyl acetate 10:1). ¹H NMR δ 2.00–2.07 (m, 6H), 6.59 (d, $J=16.8$ Hz, 1H), 6.70 (d, $J=16.8$ Hz, 1H), 7.24–7.48 (m, 5H). ¹⁹F NMR δ -56.9. IR (neat): 3020, 2920, 1640, 1600, 1490, 1440, 1340, 1220, 1110, 960, 750, 690 cm⁻¹. MS m/z $228 (0.4, M^+ + 2), 227 (7, M^+ + 1), 226 (50, M^+), 211 (100).$ HRMS (EI) m/z M⁺ calcd for C₁₃H₁₃F₃, 228.1126; found, 228.1109.

4.5.3. 4-Butyl-1-phenyl-3-trifluoromethylocta-1,3-diene (8c). Yield: 26% . Colorless oil. R_f 0.65 (hexane/ethyl acetate 10:1). ¹H NMR δ 0.89–0.97 (m, 6H), 1.31–1.50 (m, 8H), $2.20 - 2.40$ (m, 4H), 6.60 (d, $J=16.4$ Hz, 1H), 6.73 (d, J=16.4 Hz, 1H), 7.26–7.46 (m, 5H). ¹³C NMR δ 13.89, 13.93, 22.9, 23.1, 30.7, 31.1, 33.5, 34.0, 123.4 (q, J=28.0 Hz), 124.5 (q, J=275.0 Hz), 126.3, 127.8, 128.7, 132.7, 137.3, 152.5. ¹⁹F NMR δ -56.2. IR (neat): 2960, $2870, 1590, 1460, 1340, 1230, 1110, 960, 750, 690$ cm⁻¹. MS m/z 312 (2, M⁺+2), 311 (17, M⁺+1), 310 (78, M⁺), 197 (100). Anal. calcd for $C_{19}H_{25}F_3$: C, 73.52; H, 8.12. Found: C, 73.45; H, 8.24.

4.5.4. 4-Methyl-1-phenyl-3-trifluoromethylpent-3-en-1 yne (8d). Yield: 55%. Colorless oil. R_f 0.63 (hexane/ethyl acetate 10:1). ¹H NMR δ 2.08 (q, J=2.2 Hz, 3H), 2.18 (q, $J=2.2$ Hz, 3H), $7.31-7.37$ (m, 3H), $7.44-7.51$ (m, 2H). ¹⁹F NMR δ -58.4. IR (neat): 3050, 2910, 2200, 1620, 1590, 1480, 1440, 1370, 1340, 1250, 1230, 1160, 1120, 1040, 890, 750 cm^{-1} . MS m/z 226 (0.8, M⁺+2), 225 (13, M⁺+1), 224 (100, M⁺), 155 (54). HRMS (EI) m/z M⁺ calcd for $C_{13}H_{11}F_3$, 224.0813; found, 224.0802.

4.5.5. 4-Butyl-1-phenyl-3-trifluoromethyloct-3-en-1-yne (8e). Yield: 71%. Colorless oil. R_f 0.59 (hexane/ethyl acetate 10:1). ¹H NMR δ 0.90–0.99 (m, 6H), 1.32–1.63 (m, 8H), 2.36 (m, 2H), 2.49 (t, $J=7.2$ Hz, 2H), $7.32-7.37$ (m, 3H), 7.42–7.48 (m, 2H). 13C NMR ^d 13.77, 13.80, 22.9, $23.0, 30.0, 60.6, 32.5, 35.5, 82.8, 95.3, 11.3$ (q, $J=33.5$ Hz), 122.5 (g, $J=273.0$ Hz), 123.0, 128.3, 128.4, 131.3, 162.3. ¹⁹F NMR δ –57.8. IR (neat): 2960, 2870, 1610, 1590, 1490, 1460, 1350, 1260, 1160, 1120, 1060, 750, 690 cm⁻¹. MS m/ z 310 (2, M⁺+2), 309 (19, M⁺+1), 308 (94, M⁺), 196 (100). Anal. calcd for $C_{19}H_{23}F_3$: C, 74.00; H, 7.52. Found: C, 73.73; H, 7.58.

4.5.6. 1,1,1-Trifluoro-3-methyl-2-phenylbut-2-ene (8f). Yield: 69%. Colorless oil. R_f 0.68 (hexane/ethyl acetate 10:1). ¹H NMR δ 1.62 (q, J=2.4 Hz, 3H), 2.07 (q, J=2.4 Hz, 3H), 7.13–7.17 (m, 2H), 7.32–7.42 (m, 3H). ¹⁹F NMR δ -56.3 (t, J=2.6 Hz). MS m/z 202 (0.8, M⁺+2), 201 (12, M⁺+1), 200 (100, M⁺), 185 (41). HRMS (EI) m/z M^+ Calcd for $C_{11}H_{11}F_3$, 200.0813; found, 200.0821.

4.5.7. 3-Butyl-1,1,1-trifluoro-2-phenylhep-2-ene (8g). Yield: 60% . Colorless oil. R_f 0.66 (hexane/ethyl acetate 10:1). ¹H NMR δ 0.73 (t, J=7.2 Hz, 3H), 0.96 (t, J=7.2 Hz, $3H$, $1.00-1.53$ (m, $8H$), 1.88 (t, $J=6.8$ Hz, $2H$), $2.33-2.40$ (m, 2H), 7.14–7.18 (m, 2H), 7.34–7.41 (m, 3H). 13C NMR ^d 13.7, 13.9, 22.6, 23.0, 30.3, 31.1, 31.6, 33.3, 124.3 (q, $J=274.0$ Hz), 127.1 (q, $J=29.0$ Hz), 127.6, 128.2, 130.0, 135.9, 152.0 (q, J=2.7 Hz). ¹⁹F NMR δ -56.0. IR (neat): 2960, 2870, 1640, 1460, 1330, 1240, 1150, 1110, 910, 760, 700 cm^{-1} . MS m/z 286 (0.7, M⁺+2), 285 (9, M⁺+1), 284 $(5, M⁺), 172$ (100). Anal. calcd for C₁₇H₂₃F₃: C, 71.80; H, 8.15. Found: C, 71.98; H, 8.37.

4.5.8. (E)-4-Methyl-1-phenyl-3-trifluoromethyloct-3-en-**1-yne (5a).** R_f 0.60 (hexane/ethyl acetate 10: 1). ¹H NMR δ 0.96 (t, $J=7.0$ Hz, 3H), $1.26-1.62$ (m, 4H), 2.05 (q, $J=2.2$ Hz, 3H), 2.51 (dt, $J=7.7$, 1.3 Hz, 2H), 7.29–7.34 (m, 3H), $7.39-7.48$ (m, 2H). ¹³C NMR δ 13.9, 19.1 (q, $J=1.9$ Hz), 22.5, 29.5 (q, $J=1.2$ Hz), 38.17, 82.53 (q, $J=3.7 \text{ Hz}$), 95.09, 111.5 (q, $J=33.9 \text{ Hz}$), 122.5 (q, J=272.9 Hz), 123.0, 128.3, 128.4, 131.3, 157.6. ¹⁹F NMR δ -58.5. IR (neat): 3050, 2950, 2850, 2200, 1615, 1595, 1485, 1440, 1350, 1255, 1225, 1120, 1080, 1010, 910, 750, 690 cm⁻¹. MS m/z 268 (3, M⁺+2), 267 (29, M⁺+1), 266 $(100, M⁺)$, 251 (4, M⁺-Me), 197 (53, M⁺-CF₃). Anal. calcd for $C_{16}H_{17}F_3$: C, 72.16; H, 6.43. Found: C, 72.22; H, 6.51.

4.5.9. (E)-4-Methyl-1-phenyl-3-trifluoromethyloct-3-en-**1-yne (5b).** R_f 0.58 (hexane/ethyl acetate 10:1). ¹H NMR δ 0.95 (t, $J=7.0$ Hz, 3H), 1.10 (t, $J=7.5$ Hz, 3H), 1.36–1.62 (m, 4H), 2.38 (ddq, J=7.5, 1.6, 1.6 Hz, 2H), 2.45–2.53 (m, 2H), 7.28–7.35 (m, 3H), 7.38–7.48 (m, 2H). 13C NMR ^d 12.9, 13.9, 22.9, 25.8 (q, J=1.9 Hz), 30.0, 35.1, 82.8 (q, $J=3.0$ Hz), 95.2, 111.1 (q, $J=33.8$ Hz), 122.5 (q, J = 272.9 Hz), 123.0, 128.35, 128.41, 131.3, 163.4 (q, $J=2.3$ Hz). ¹⁹F NMR δ -58.0. IR (neat): 3057, 2959, 2876, 2210, 1612, 1595, 1491, 1458, 1445, 1356, 1275, 1256, 1217, 1165, 1123, 1055, 999, 914, 754, 729, 689 cm⁻¹. MS m/z 282 (3, M⁺+2), 281 (26, M⁺+1), 280 $(100, M⁺), 223 (38, M⁺-Bu), 251 (27, M⁺-Et), 211 (9,$ M^+ – CF₃). Anal. calcd for C₁₆H₁₇F₃: C, 72.84; H, 6.83. Found: C, 72.80; H, 6.90.

4.6. Typical procedure for the preparation of tetrasubstituted alkenes 5

To a solution of $1c(280 \text{ mg}, 1.0 \text{ mmol})$ in THF (5 mL) was added BuLi (1.50 M, 2.0 mL, 3.0 mmol) at -98° C. The solution was stirred at -98° C for 15 min and then was added PhC \equiv CB(OCMe₂)₂ (460 mg, 2.0 mmol) at -98° C. The resulting solution was gradually warmed to room temperature and then refluxed for 4 h before quenching with aqueous saturated $NH₄Cl$ solution. The aqueous layer was extracted with diethyl ether three times. The combined

organic layer was dried over anhydrous $MgSO₄$ and concentrated in vacuo. The subsequent purification by column chromatography on silica gel (hexane only) gave 5c (260 mg, 74% yield).

4.6.1. (Z)-1-Phenyl-4-phenylethynyl-3-trifluoromethyloct-3-en-1-yne (5c). Pale yellow oil. R_f 0.58 (hexane/ethyl acetate 10: 1). ¹H NMR δ 0.99 (t, J=7.1 Hz, 3H), 1.42 (tq, $J=7.2$, 7.1 Hz, 2H), 1.74 (tt, $J=7.4$, 7.2 Hz, 2H), 2.70 (t, $J=7.4$ Hz, 2H), $7.32-7.41$ (m, 6H). ¹³C NMR δ 13.9, 22.2, 30.0, 36.2, 82.6, 86.3, 101.7, 104.2, 118.5 (q, J=33.4 Hz), 121.6 (g, $J=273.1$ Hz), 122.4, 128.4 (2C), 129.1, 131.4 (2C), 131.9, 137.9. ¹⁹F NMR δ -61.1. IR (neat): 3063, 2959, 2930, 2862, 2210, 2187, 1597, 1570, 1491, 1443, 1356, 1242, 1192, 1132, 1069, 914, 754, 689, 665 cm⁻¹. MS m/z 354 (4, M⁺+2), 353 (28, M⁺+1), 352 (100, M⁺), 241 (37, M⁺-PhCC). Anal. calcd for C₂₃H₁₉F₃: C, 78.39; H, 5.43. Found: C, 78.63; H, 5.36.

4.6.2. (3Z,5E)-4-Butyl-1-phenyl-3-trifluoromethyldec-**3,5-dien-1-yne (5d).** Yield: 56%. Pale yellow oil. R_f 0.60 (hexane/ethyl acetate 10:1). ¹H NMR δ 0.92 (t, J=7.4 Hz, 3H), 0.96 (t, $J=7.0$ Hz, 3H), $1.27-1.60$ (m, 8H), 2.21 (t, $J=7.8$ Hz, 2H), 2.67 (t, $J=7.5$ Hz, 2H), 6.17 (dt, $J=15.6$, 7.8 Hz, 1H), 6.59 (d, $J=15.6$ Hz, 1H), 7.28–7.34 (m, 3H), 7.39–7.50 (m, 2H). 13C NMR ^d 13.9, 14.0, 22.3, 23.0, 31.1, 31.3, 31.8, 33.3, 84.1, 98.3, 110.1 (q, J=33.6 Hz), 122.5 (q, J = 273.5 Hz), 123.1, 125.5, 128.4, 128.5, 131.3, 139.5, 154.0. ¹⁹F NMR δ -56.0. IR (neat): 3063, 2961, 2934, 2874, 2203, 1784, 1684, 1597, 1491, 1469, 1350, 1242, 1171, 1126, 1069, 908, 756, 733, 689, 665 cm⁻¹. MS m/z $336 (0.2, M^{+}+2), 335 (3, M^{+}+1), 334 (14, M^{+}), 236 (100).$ HRMS (EI) m/z M⁺ calcd for C₂₁H₂₅F₃, 334.1908; found, 334.1877.

4.6.3. (Z)-1,4-Diphenyl-3-trifluoromethyloct-3-en-1-yne (5e). Yield: 66% as colorless crystals, mp 67°C. R_f 0.59 (hexane/ethyl acetate 10:1). ¹H NMR δ 0.89 (t, J=7.0 Hz, 3H), $1.27-1.44$ (m, 4H), 2.80 (t, $J=6.4$ Hz, 2H), $7.10-$ 7.21 (m, 2H), 7.27–7.40 (m, 6H), 7.46–7.53 (m, 2H). ¹³C NMR δ 13.9, 22.5, 29.2, 39.1, 82.7, 95.9, 112.2 (q, J=32.8 Hz), 121.9 (q, J=273.1 Hz), 122.7, 127.0, 127.9, 128.0, 128.4, 128.8, 131.5, 138.8, 158.9. 19F NMR ^d 257.2. IR (nujol): 2926, 2858, 2210, 1589, 1572, 1489, 1445, 1344, 1231, 1182, 1096, 1070, 1011, 920, 779, 756, 706, 691, 646 cm⁻¹. MS m/z 330 (3, M⁺+2), 329 (22, $M^{+}+1$), 328 (100, M^{+}), 271 (15, $M^{+}-Bu$), 251 (15, M^+ –Ph). HRMS (EI) M^+ calcd for C₂₁H₁₉F₃, 328.1439; found, 328.1471.

4.6.4. (E)-1-4-Diphenyl-3-trifluoromethyloct-3-en-1-yne (5e'). Yield: 84%. Colorless oil. R_f 0.59 (hexane/ethyl acetate 10:1). ¹H NMR δ 0.85 (t, J=6.8 Hz, 3H), 1.21-1.34 $(m, 4H), 2.64 - 2.80$ $(m, 2H), 7.09 - 7.58$ $(m, 10H)$. ¹³C NMR δ 13.7, 22.6, 30.4, 34.5, (q, J=1.9 Hz), 93.8, 94.4 112.5 (q, $J=33.6$ Hz), 122.5 (q, $J=273.5$ Hz), 122.7, 128.0, 128.2, 128.3, 128.4, 128.7, 131.2, 140.3, 159.7. ¹⁹F NMR δ – 58.0. IR (neat): 3059, 2961, 2932, 2874, 2212, 1591, 1491, 1442, 1356, 1271, 1248, 1171, 1126, 1069, 914, 756, 691, 665, 646 cm⁻¹. MS m/z 330 (3, M⁺+2), 329 (25, M⁺+1), 328 $(100, M⁺), 271 (14, M⁺-Bu), 251 (14, M⁺-Ph). HRMS$ (EI) m/z M⁺ calcd for C₂₁H₁₉F₃, 328.1439; found, 328.1425.

4.6.5. (E)-2-(1-Butyl-4-phenyl-2-trifluoromethylbut-1 en-3-ynyl)-4,4,5,5-tetramethyl-1,3,2-dioxaborolane (6E, \mathbb{R}^2 =Bu). Yield: 76%. Pale yellow oil. R_f 0.34 (hexane/ ethyl acetate 10:1). ¹H NMR δ 0.94 (t, J=7.0 Hz, 3H), 1.31 $(s, 12H), 1.35-1.62$ (m, 4H), 2.57 (t, J=6.8 Hz, 2H), 7.24– 7.36 (m, 3H), $7.42 - 7.49$ (m, 2H). ¹³C NMR δ 13.8, 22.6, 24.6, 30.2, 34.1, 81.5, 84.5, 96.9, 119.3 (q, $J=33.8$ Hz), 122.2 (q, J=271.9 Hz), 122.4, 128.3, 128.8, 131.6, 151.0. ¹⁹F NMR δ –62.9. IR (neat): 2961, 2861, 2208, 1612, 1491, 1445, 1346, 1249, 1180, 1134, 1030, 962, 851, 756, 689, 665 cm⁻¹. MS m/z 380 (3, M⁺+2), 379 (25, M⁺+1), 378 (100, M⁺), 377 (23, M⁺-1). Anal. calcd for C₂₁H₂₆BF₃O₂: C, 66.68; H, 6.93. Found: C, 66.44; H, 6.96.

4.6.6. (E)-2-(1,4-Diphenyl-2-trifluoromethyl-but-1-en-3 ynyl)-4,4,5,5-tetramethyl-[1,3,2]dioxaborolane (6E, R^2 =Ph). Yield: 86%. Pale yellow oil. R_f 0.24 (hexane/ ethyl acetate 10:1). ¹H NMR (500 MHz) δ 1.29 (s, 12H), 7.24–7.40 (m, 8H), 7.58 (m, 2H). ¹³C NMR (125 MHz) δ 24.6, 82.8, 84.8, 95.8, 119.2 (q, J=33.5 Hz), 122.2, 123.5 (q, J = 272.2 Hz), 128.1, 128.3 (2C), 128.6, 128.9, 131.6, 137.5, 147.7. ¹⁹F NMR δ -63.1. IR (neat): 3058, 2981, 2931, 2208, 1737, 1602, 1490, 1444, 1373, 1340, 1253, 1217, 1137, 1068, 977, 850 cm⁻¹. MS m/z 400 (1, M⁺+2), 399 (7, M^{+} +1), 398 (28, M⁺), 298 (100). Anal. calcd for $C_{23}H_{22}BF_3O_2$: C, 69.37; H, 5.57. Found: C, 69.63; H, 5.83.

4.6.7. (Z)-2-(1-Butyl-4-phenyl-2-trifluoromethyl-but-1 en-3-ynyl)-4,4,5,5-tetramethyl-1,3,2-dioxaborolane (6Z, \mathbb{R}^2 =Bu). Yield: 54%. Pale yellow oil. R_f 0.34 (hexane/ ethyl acetate 10:1). ¹H NMR δ 0.95 (t, J=6.8 Hz, 3H), 1.32 (s, 12H), 1.17–1.63 (m, 4H), 2.41–2.59 (m, 2H), 7.29–7.34 (m, 2H), $7.43 - 7.48$ (m, 3H), ¹³C NMR δ 13.8, 22.7, 24.7, 31.2, 31.6, (q, $J=1.5$ Hz), 84.1, 84.4, 93.0, 120.0 (q, $J=30.0$ Hz), 121.6 (q, $J=276.0$ Hz), 122.7, 128.3, 128.6, 131.5, 151.4. ¹⁹F NMR δ -59.2. IR (neat): 2980, 2931, 2874, 2210, 1599, 1490, 1445, 1381, 1310, 1246, 1178, 1063, 964, 847, 756, 691, 667 cm⁻¹. MS m/z 380 (3, M^{+} +2), 379 (23, M^{+} +1), 378 (100, M^{+}), 377 (23, M^{+} -1). HRMS (EI) m/z M⁺ calcd for C₂₁H₂₆BF₃O₂, 378.1978; found, 378.2009.

4.6.8. (Z)-2-(1,4-Diphenyl-2-trifluoromethyl-but-1-en-3 ynyl)-4,4,5,5-tetramethyl-[1,3,2]dioxaborolane (6Z, $\mathbf{R}^2 = \mathbf{Ph}$). Yield: 31%. Pale yellow oil. R_f 0.24 (hexane/ ethyl acetate 10:1). ¹H NMR δ 1.28 (s, 12H), 7.22–7.55 (m, 10H). ¹³C NMR δ 24.4, 84.2, 84.8, 94.2, 120.1 (q, J=33.4 Hz), 121.1 (q, J=275.7 Hz), 122.3, 127.1, 127.8, 128.1, 128.4, 129.0, 131.7, 136.9, 150.1. ¹⁹F NMR δ -58.5. MS m/z 400 (0.6, M⁺+2), 399 (4, M⁺+1), 398 (14, M⁺), 298 (100). Anal. calcd for C₂₃H₂₂BF₃O₂: C, 69.37; H, 5.57. Found: C, 69.20; H, 5.45.

4.6.9. (Z)-(1,4-Diphenyl-2-trifluoromethyl-but-1-en-3 ynyl)-dimethyl-phenyl-silane (9). Yield: 14%. Pale yellow oil. R_f 0.46 (hexane/ethyl acetate 10:1). ¹H NMR δ 0.33 (s, 3H), 0.34 (s, 3H), 6.96–7.04 (m, 3H), 7.17–7.38 (m, 10H), 7.48–7.56 (m, 2H). ¹³C NMR δ -0.77, 0.85, 84.0, 98.3, 120.8 (q, $J=272.8$ Hz), 122.2, 125.7 (q, $J=35.3$ Hz), 126.5, 126.7, 127.68, 127.73, 128.0, 128.1, 128.8, 129.2, 131.6, 133.0, 133.8, 136.9, 139.8, 143.4. ¹⁹F NMR δ -60.1. IR (neat): 3069, 3053, 3022, 2956, 2204, 1596, 1489, 1427, 1338, 1249, 1168, 1132, 1066, 939, 813, 756, 700 cm⁻¹.

MS m/z 408 (7, M⁺+2), 407 (27, M⁺+1), 406 (78, M⁺), 309 (100). HRMS (EI) m/z M⁺ calcd for C₂₅H₂₁F₃Si, 406.1365; found, 406.1362.

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